© 2004 International Union of Crystallography Printed in Great Britain - all rights reserved

Acta Crystallographica Section E

Sema Öztürk,<sup>a</sup> Mehmet

Orhan Büyükgüngör<sup>c</sup> and

Akkurt,<sup>a</sup>\* Edmont Stoyanov,<sup>b</sup>

<sup>a</sup>Department of Physics, Faculty of Arts and

Sciences, Erciyes University, 38039 Kayseri,

Turkey, <sup>b</sup>Faculty of Pharmacy, Medical

University of Sofia, Dunav 2, Sofia 1000, Bulgaria, <sup>c</sup>Department of Physics, Ondokuz

Mayıs University, 55139 Samsun, Turkey, and <sup>d</sup>Pharmaceutical Institute, University of Kiel, Gutenbergstraße 76, D-24118 Kiel, Germany Correspondence e-mail: akkurt@erciyes.edu.tr

**Structure Reports** 

Online

ISSN 1600-5368

Dieter Heber<sup>d</sup>

Key indicators

R factor = 0.038 wR factor = 0.108

T = 293 K

Single-crystal X-ray study

Mean  $\sigma(C-C) = 0.002$  Å

http://journals.iucr.org/e.

Data-to-parameter ratio = 13.6

For details of how these key indicators were

automatically derived from the article, see

Methyl 6-amino-5-cyano-2-methoxycarbonylmethyl-4-phenyl-4H-pyran-3-carboxylate

The pyran ring in the title compound,  $C_{17}H_{16}N_2O_5$ , adopts a boat conformation. The crystal packing is stabilized by intermolecular N-H···O and N-H···N hydrogen bonds.

# Comment

A series of 4H-pyran and cyclohexanone derivatives was prepared via the three-component reaction of dimethyl acetonedicarboxylate, aromatic aldehydes and malononitrile. We present here the structure of the title compound, (I).

The molecular structure of (I) is shown in Fig. 1. The O1-C1 and O1-C5 bond lengths are 1.387 (2) and 1.363 (2) Å, respectively (Table 1). The mean C=O bond length of 1.198 (2) Å is typical of a double bond, and is consistent with a previously published structure (Nesterov & Viltchinskaia, 2001). The phenyl ring system is planar within 0.003 (1) Å. The pyran ring adopts a boat conformation, with atoms O1 and C3 deviating from the C1/C2/C4/C5 plane by 0.079 (1) and 0.111 (1) Å, respectively (Cremer & Pople, 1975).

The crystal packing (Fig. 2) is stabilized by intermolecular  $N-H \cdots O$  and  $N-H \cdots N$  hydrogen bonds (Table 2).

# **Experimental**

Compound (I) was synthesized according to the method of Heber & Stoyanov (2003).

Figure 1



N2





Received 15 January 2004 Accepted 20 January 2004 Online 30 January 2004

Crystal data

# $\begin{array}{l} C_{17}H_{16}N_2O_5\\ M_r = 328.32\\ \text{Triclinic, } P\overline{1}\\ a = 8.1635\ (8)\ \text{\AA}\\ b = 10.0366\ (9)\ \text{\AA}\\ c = 10.6641\ (10)\ \text{\AA}\\ \alpha = 67.340\ (7)^\circ\\ \beta = 87.008\ (8)^\circ\\ \gamma = 76.740\ (7)^\circ\\ V = 784.17\ (14)\ \text{\AA}^3 \end{array}$

### Data collection

Stoe IPDS-II diffractometer  $\omega$  scans Absorption correction: by integration (*X-RED32*; Stoe & Cie, 2002)  $T_{min} = 0.968, T_{max} = 0.981$ 11291 measured reflections

### Refinement

Refinement on  $F^2$  $R[F^2 > 2\sigma(F^2)] = 0.038$  $wR(F^2) = 0.108$ S = 1.053065 reflections 225 parameters

# Table 1

Selected geometric parameters (Å, °).

O1-C1	1.3874 (18)	O4-C16	1.317 (2)
O1-C5	1.3631 (18)	O4-C17	1.444 (2)
O2-C13	1.202 (2)	O5-C16	1.193 (2)
O3-C13	1.326 (2)	N1-C6	1.145 (2)
O3-C14	1.444 (2)	N2-C5	1.336 (2)
C1-O1-C5	120.16 (11)	N1-C6-C4	177.64 (17)
C13-O3-C14	117.52 (13)	O3-C13-C2	115.45 (13)
C16-O4-C17	117.07 (14)	O2-C13-O3	122.24 (15)
O1-C1-C15	106.70 (12)	O2-C13-C2	122.32 (15)
O1-C1-C2	122.21 (14)	O4-C16-O5	124.03 (15)
O1-C5-N2	110.40 (13)	O4-C16-C15	113.32 (14)
O1-C5-C4	121.13 (13)	O5-C16-C15	122.65 (15)
N2-C5-C4	128.47 (15)		
C1-O1-C5-N2	173.34 (12)	C1-C2-C13-O3	0.1 (2)
C14-O3-C13-C2	176.63 (13)	C3-C2-C13-O2	-2.4(2)
C14-O3-C13-O2	-3.6(2)	C1-C2-C13-O2	-179.64(15)
C17-O4-C16-O5	1.4 (3)	C6-C4-C5-N2	-0.1(2)
C17-O4-C16-C15	-177.66(15)	C3-C4-C5-N2	172.46 (14)
C2-C1-C15-C16	94.67 (19)	C1-C15-C16-O5	163.51 (16)
O1-C1-C15-C16	-81.92 (15)	C1-C15-C16-O4	-17.38 (19)

# Table 2

Hydrogen-bonding geometry (Å, °).

$\overline{D - \mathbf{H} \cdots A}$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
$\begin{array}{c} N2 - H2A \cdots O2^{i} \\ N2 - H2B \cdots N1^{ii} \end{array}$	0.853 (19)	2.186 (19)	3.038 (2)	176.8 (15)
	0.89 (2)	2.24 (2)	3.132 (2)	177.8 (16)

Symmetry codes: (i) x - 1, y, z; (ii) 1 - x, -y, 2 - z.

Z = 2  $D_x = 1.390 \text{ Mg m}^{-3}$ Mo K $\alpha$  radiation Cell parameters from 8916 reflections  $\theta = 2.1-27.0^{\circ}$   $\mu = 0.10 \text{ mm}^{-1}$  T = 293 KPrism, colorless  $0.40 \times 0.33 \times 0.27 \text{ mm}$ 

3065 independent reflections 2265 reflections with  $I > 2\sigma(I)$  $R_{int} = 0.024$  $\theta_{max} = 26.0^{\circ}$  $h = -10 \rightarrow 10$  $k = -12 \rightarrow 12$  $l = -13 \rightarrow 13$ 

H atoms treated by a mixture of independent and constrained refinement  $w = 1/[\sigma^2(F_o^2) + (0.0681P)^2]$ where  $P = (F_o^2 + 2F_c^2)/3$   $(\Delta/\sigma)_{\rm max} < 0.001$   $\Delta\rho_{\rm max} = 0.21$  e Å<sup>-3</sup>  $\Delta\rho_{\rm min} = -0.17$  e Å<sup>-3</sup>



### Figure 2

The crystal packing and hydrogen bonding (dashed lines) in (I). [Symmetry codes: (i) x - 1, y, z; (ii) 1 - x, -y, 2 - z; (iii) 1 + x, y, z.]

The C-bound H atoms were positioned geometrically and were treated as riding on their parent C atoms, with C–H distances in the range 0.93–0.97 Å and  $U_{\rm iso}({\rm H}) = 1.5 U_{\rm eq}({\rm C})$  for methyl H atoms and  $1.2 U_{\rm eq}({\rm C})$  for aromatic H atoms. The amino H atoms, H2A and H2B, were located in a difference Fourier map and their positional and isotropic displacement parameters were refined.

Data collection: X-AREA (Stoe & Cie, 2002); cell refinement: X-AREA; data reduction: X-RED32 (Stoe & Cie, 2002); program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP-3 (Farrugia, 1997); software used to prepare material for publication: WinGX (Farrugia, 1999).

The authors acknowledge the Faculty of Arts and Sciences, Ondokuz Mayis University, Turkey, for the use of the Stoe IPDS-II diffractometer (purchased under grant No. F.279 of the University Research Fund).

# References

- Cremer, D. & Pople, J. A. (1975). J. Am. Chem. Soc. 97, 1354-1358.
- Farrugia, L. J. (1997). J. Appl. Cryst. 30, 565.
- Farrugia, L. J. (1999). J. Appl. Cryst. 32, 837-838.
- Heber, D. & Stoyanov, E. V. (2003). Synthesis, 2, 227-232.
- Nesterov, V. N. & Viltchinskaia, E. A. (2001). Acta Cryst. C57, 616-618.
- Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
- Stoe & Cie (2002). X-AREA (Version 1.18) and X-RED32 (Version 1.04). Stoe & Cie, Darmstadt, Germany.